

The influence of base-pair tautomerism on single point mutation rates in aqueous DNA

Alex Gheorghiu [1], **Alya A. Arabi** [2], **Peter V. Coveney** [1],[3]

[1] *Centre for Computational Chemistry, Department of Chemistry, University College London, UK*

[2] *College of Natural and Health Science, Zayed University, Abu Dhabi, UAE*

[3] *Informatics Institute, University of Amsterdam, Netherlands*

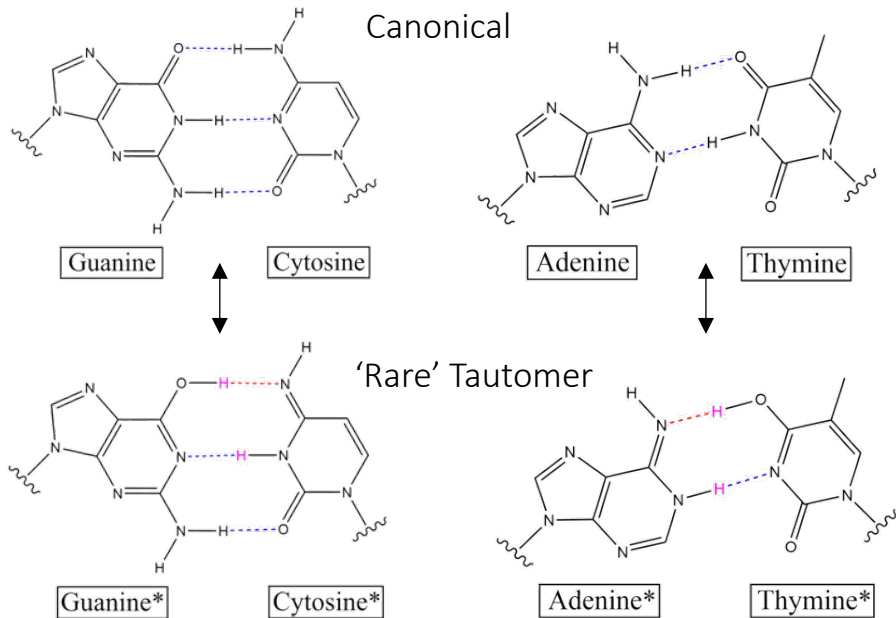
Mutations in DNA

- External agents such as external electric fields, or other carcinogenic compounds are known to facilitate DNA mutations^[1]
- ... But what about *spontaneous mutations*?

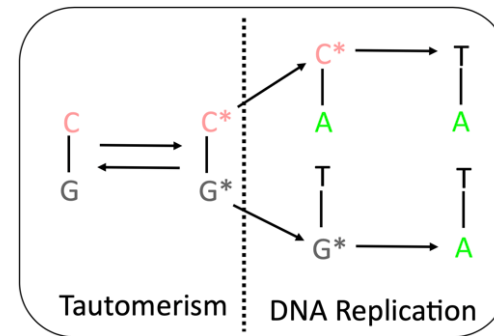


(1) Loft, Steffen, and Henrik E. Poulsen. "Cancer risk and oxidative DNA damage in man." *Journal of molecular medicine* 74, no. 6 (1996): 297-312.

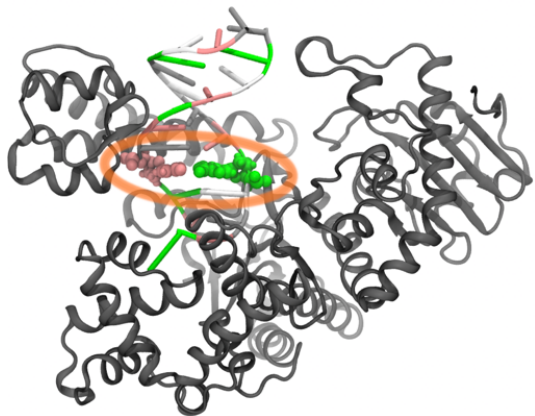
Watson-Crick Base-Pairs and Löwdin's Hypothesis



- Protons obey the laws of quantum theory, behaving like 'wave packets'.
- Due to the quantum-mechanical tunnelling effect, there is always a small probability of proton transfer within any given base-pair.



'...this transfer of proton over distances less than 1 Å may be the driving force for genetic mutations in all living organisms...' – Löwdin^[2]



- DNA polymerase caught red-handed!
- Structural evidence of a C:A mismatch within crystallised DNA polymerase
- An overall **G:C** → **A:T** mutation bias has been reported in a variety of organisms

QM-only gas-phase studies

- Proton transfer in G:C is energetically more favourable than in A:T^[3]
- Double proton transfer; step-wise or concerted mechanism? ^[4]
- Little to no quantification of errors, due to lack of replicas
- A gas-phase base-pair is an idealised structure – is it truly representative?
- Provide a good starting point

[3] Jacquemin, Denis, et al. "Assessing the importance of proton transfer reactions in DNA." *Accounts of chemical research* 47.8 (2014): 2467-2474.

[4] Sekiya, H., & Sakota, K. (2008). Excited-state double-proton transfer in a model DNA base pair: Resolution for stepwise and concerted mechanism controversy... *Journal of Photochemistry and Photobiology C: Photochemistry Reviews*, 9(2), 81-91.

Considerations when modelling aqueous DNA

Reproducibility?

A realistic model; solvated, large enough DNA sample, >10,000 atoms

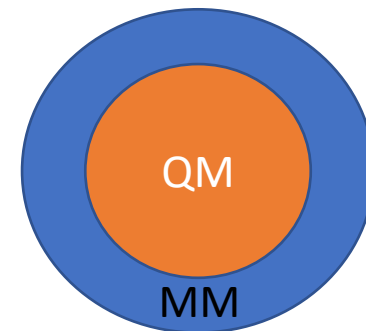
Proton transfer is both a bond breaking/forming effect

DNA is a large, complex biomolecule molecule

We can utilise both of these methods simultaneously^[5]: **(QM/MM)**

Requires quantum mechanics (**QM**) to model: Expensive at high accuracy

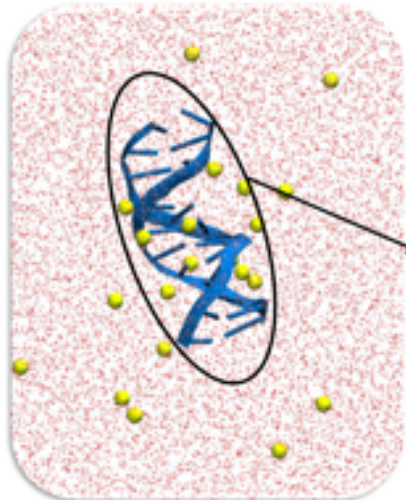
Use forcefield molecular mechanics (**MM**): Computationally cheaper



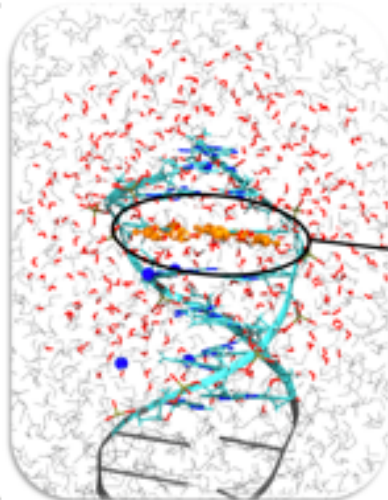
[5] Lu, You, Zhenggang Lan, and Walter Thiel. "Hydrogen bonding regulates the monomeric nonradiative decay of adenine in DNA strands." *Angewandte Chemie International Edition* 50.30 (2011): 6864-6867.

Modelling Proton Transfer: Our Multiscale Workflow

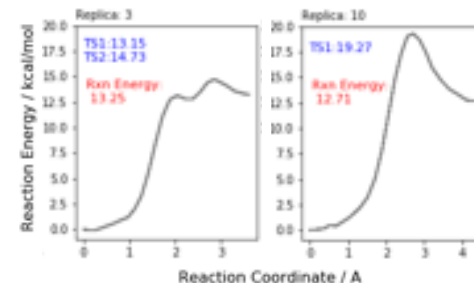
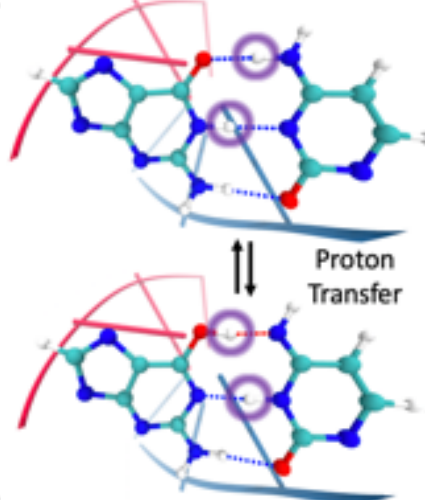
a) Ensemble MD



b) Ensemble QM/MM



c) QM Region



25 QM/MM reaction pathways mapped with nudged elastic band per base-pair

a) MD Code: NAMD
AMBER *parmbsc1* force field at 300 K, 1 atm. **10** replicas of 10 ns simulation to thermalize DNA effectively.

b) QM/MM Code: ChemShell
Linking NWChem (QM) with DL-POLY (MM). Ensemble QM/MM, using configurations drawn from MD*. **25** replicas for a given base-pair.

c) QM Code: NWChem
One base pair (~30 atoms).
QM approximation:
B3LYP+XDM/aug-cc-pvdz

* Snapshots are chosen on an average base-pair distance criteria

Advantages to using Ensemble QM/MM

- Inclusion of explicit solvation and DNA structure - more realistic than gas-phase!
- Observe multiple reaction paths for the same base-pair
- Determine the probability of each pathway occurring
- Comment on the statistical variance of the rates for each pathway

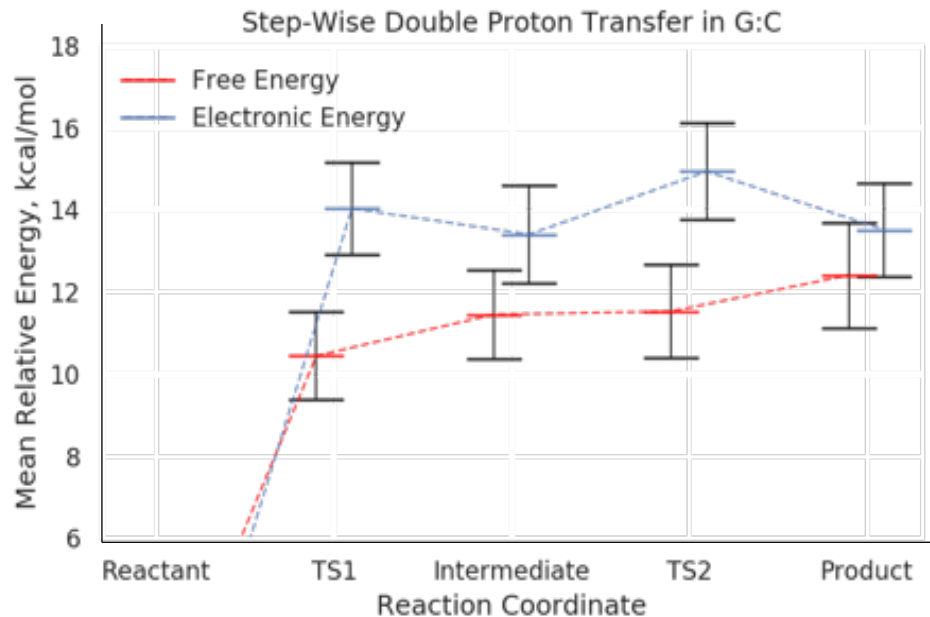
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Base-Pair	No proton transfer	SPT Concerted	DPT	
			Step-Wise	Concerted
% of G:C replicas	0	4	80	16
% of A:T replicas	56	36	0	8

Step-Wise Double Proton Transfer in G:C

- The most probable type of proton transfer in G:C is step-wise, 80% of the time
- Step-wise, two transition states and an intermediate

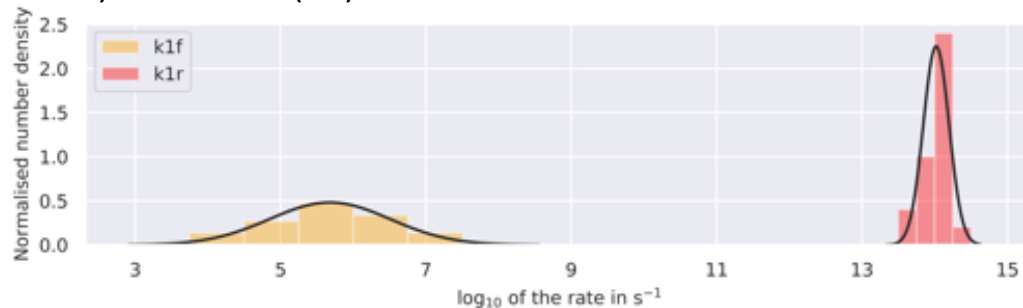


The reaction coordinate for the step-wise double proton transfer mechanism in the G:C base-pair. Error bars are standard deviation. Data points calculated from 20 replicas out of 25 total

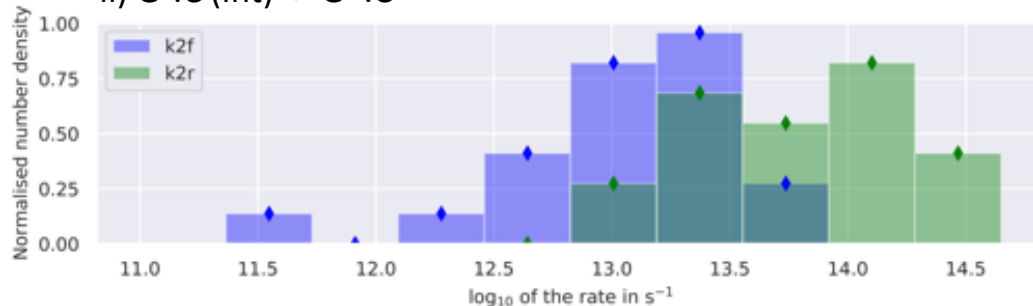
Step-Wise Double Proton Transfer in G:C

- Rate determining step is k_{1f} , many orders of magnitude slower than k_{1r}
- The use of ensemble QM/MM gives insight to the statistical variance
- The second step is much closer to equilibrium than the first
- Tautomer has a lifetime of ~ 50 fs ($1/k_{1r}$)

i) G:C \rightarrow G⁺:C(Int)



ii) G⁺:C(Int) \rightarrow G^{*}:C^{*}



The normalized histogram of log proton transfer rates for the step-wise double proton transfer mechanism in the G:C base-pair

Step-Wise Double Proton Transfer in G:C

$$G_{taut} = \left(\sum_{K_{DPT}} K_{eq,weighted} \right) \times G \quad (1)$$

- Assuming;
 - Genome has 100% G:C content
 - Every tautomerism leads to mismatch
- $G_{taut} = 0 - 114$ base-pairs per-genome
- 1 in 3.8×10^{-8} mutations per nucleotide site.
- This value is within the estimation of mutation rate per nucleotide in humans ($10^{-11}/10^{-8}$) [6]

where;

G is size of genome in base-pairs (3×10^9)

The probability of step-wise double proton transfer is 0.8

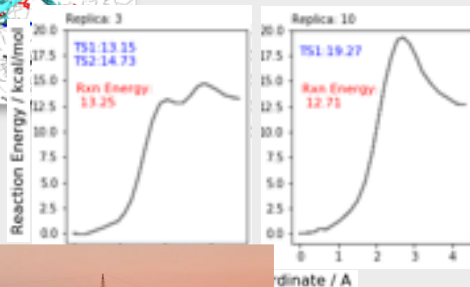
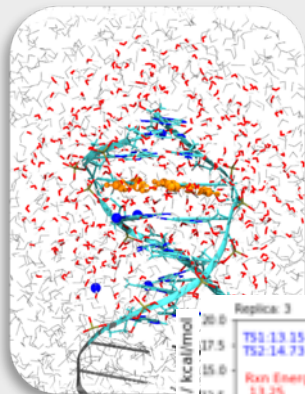
K_{eq} is the calculated average equilibrium constant (1.54×10^{-8})

$K_{eq,weighted}$ is K_{eq} multiplied by it's probability of occurrence

G_{taut} is the number of tautomeric base-pairs in the genome at any given time

[6] Nachman, Michael W., and Susan L. Crowell. "Estimate of the mutation rate per nucleotide in humans." *Genetics* 156.1 (2000): 297-304.

Conclusions

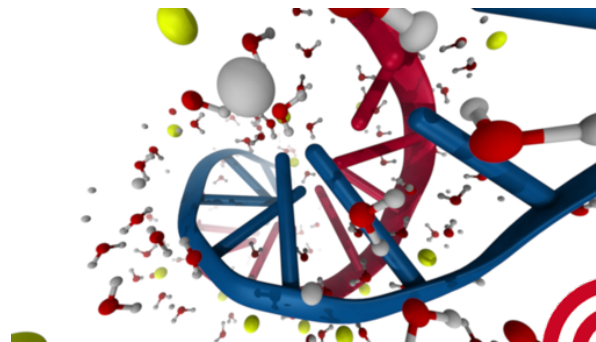


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- Proton transfer in biological systems is very sensitive to its environment, necessitating multiscale modelling
- Multiple reaction pathways for each base-pair are observed, with varying probabilities
 - G:C double proton transfer 96% of the time
 - A:T single proton transfer 36% of the time
- The frequency of a tautomer in human genome is $< 0.001\%$
- The life-times of the tautomers are fs , while DNA unwinding occurs on the ns scale

Contact: uccaagh@ucl.ac.uk

Thanks for listening!



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